

Determination of B/Ca of Natural Carbonates by HR-ICP-MS

SUPPLEMENT

Supplement 1. Calibration Standards for Metal/Ca ratio determination

Element ratios	Li/Ca $\mu\text{mol/mol}$	B/Ca $\mu\text{mol/mol}$	Na/Ca mmol/mol	Mg/Ca mmol/mol	Al/Ca mmol/mol	Mn/Ca $\mu\text{mol/mol}$	Fe/Ca mmol/mol	Zn/Ca $\mu\text{mol/mol}$	Sr/Ca mmol/mol	Cd/Ca $\mu\text{mol/mol}$	Ba/Ca $\mu\text{mol/mol}$	U/Ca nmol/mol
Standards												
Standard Solution 0	0.3	0.7	0.007	0.003	0.001	0.2	0.000	0.77	0.005	0.005	0.05	0.1
Standard Solution 1	2.2	15.5	1.970	0.199	0.020	9.9	0.010	1.80	0.478	0.025	2.00	1.0
Standard Solution 2	4.5	32.1	3.133	0.625	0.042	26.2	0.031	2.95	0.832	0.057	3.16	5.2
Standard Solution 3	6.5	62.7	4.118	1.230	0.062	51.4	0.051	4.01	1.023	0.108	4.13	15.4
Standard Solution 4	7.9	94.6	4.673	1.692	0.085	93.5	0.065	5.68	1.122	0.146	4.72	46.8
Standard Solution 5	10.5	132.7	5.693	2.271	0.114	188.1	0.084	8.59	1.317	0.193	5.69	94.4
Standard Solution 6	14.2	179.4	6.967	2.987	0.148	342.6	0.108	11.11	1.582	0.253	8.94	198.2
Standard Solution 7	18.4	228.8	8.503	4.253	0.213	529.9	0.136	17.21	1.902	0.325	12.42	367.6
Standard Solution 8	21.4	274.7	9.493	5.269	0.264	681.1	0.156	22.49	2.093	0.427	15.72	530.3

Supplement 2. B/Ca blank / contamination calculations

B/Ca ratio change for a 0.4 pg B blank during analysis:

$$[Ca]_{Matrix} : 10 \text{ ppm} = 10 \mu\text{g/ml}$$

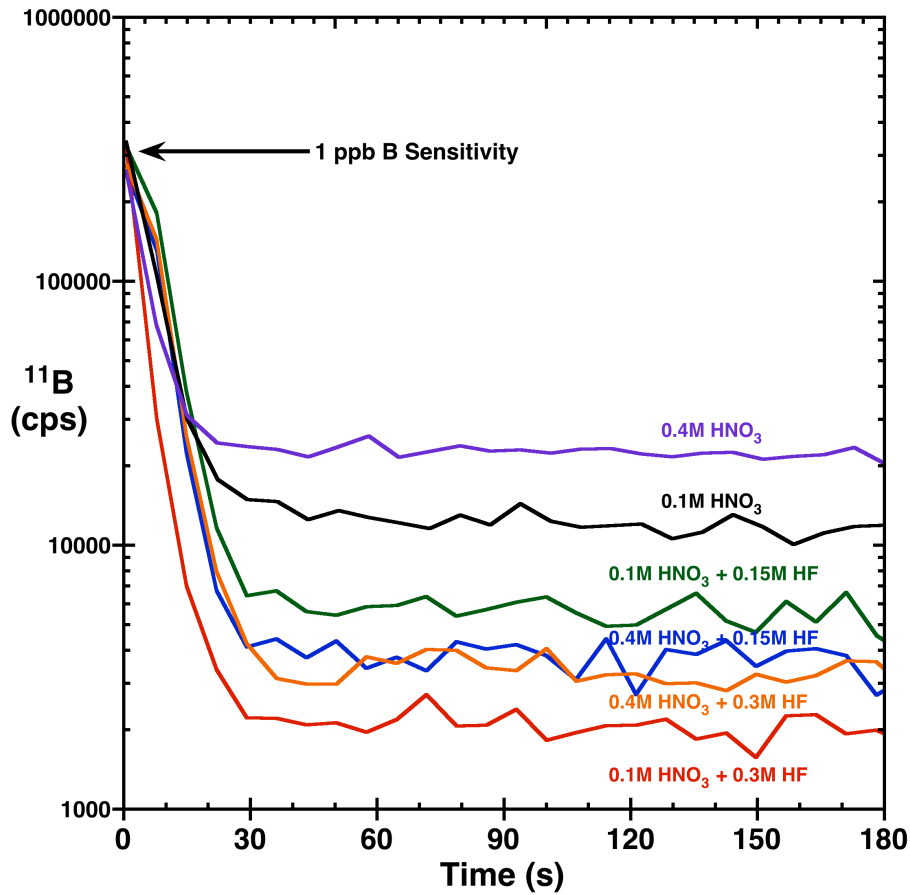
$$\text{Volume of solution per analysis} = 150 \mu\text{l} = 0.15 \text{ ml}$$

$$\text{Total Ca} = 10 \mu\text{g/ml} \times 0.15 \text{ ml} = 1.5 \mu\text{g}$$

$$\text{Mass of B contamination} = 0.4 \text{ pg}$$

$$(B/Ca)_{Contamination} = 0.4 \text{ pg} / 1.5 \mu\text{g} = 0.4 \mu\text{g} / 1.5 \text{ g} = 10.67 \mu\text{g} / 40 \text{ g}$$

$$= 1 \mu\text{mol} / 1 \text{ mol} = 1 \mu\text{mol} / \text{mol}$$

21 **Supplement 3. Boron washout in different acid matrices**

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23 **Supplement 3.** Time-resolved comparison of boron washout in different molarity HNO₃ and
 24 HF acid matrices under identical instrument conditions. Boron sensitivity is plotted on a
 25 logarithmic scale to amplify the differences between the matrices. The washout curves are
 26 color coded according to their matrix acid and matrix strength. The washout patterns were
 27 generated after aspiration of a 1 ppb (ng/ml) high purity boron solution for 180s, typical
 28 sample analysis time for $\delta^{11}\text{B}$ determination, in the same matrix under investigation.

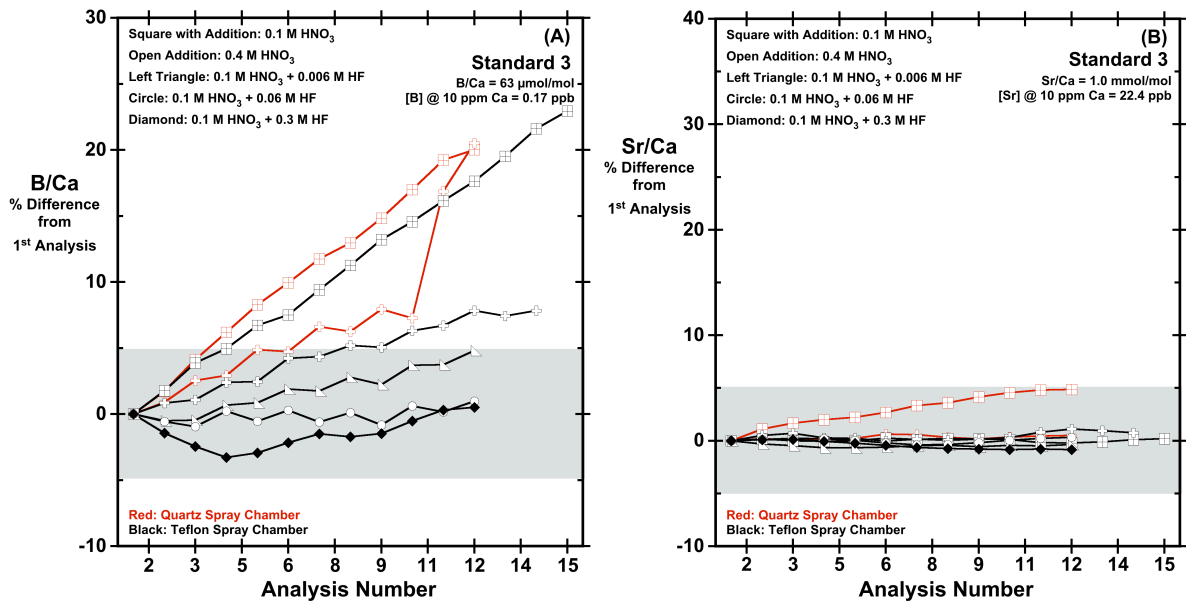
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30 **Quantitative Washout:**

31 When the post sample/standard analysis matrix boron blanks are within $\pm 1\%$ of the pre
 32 analysis matrix boron blanks (uncorrected for instrumental drift) – is defined here as
 33 ‘quantitative washout’.

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35 Supplement 4 (A) and (B). Matrix dependent B/Ca and Sr/Ca washouts



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37 Supplement 4 (A) and (B). Comparison of B/Ca and Sr/Ca memory effects in different acid
 38 matrices. Figures (A) and (B) represent back to back analyses of an intermediate
 39 concentration standard (Standard 3) bracketed by acid blanks (**Figure 1 and Supplement 1**)
 40 – the data for the Standard 3 are shown here. The figures represent percent difference
 41 between the first analysis and subsequent analyses for both B/Ca and Sr/Ca. All analyses
 42 were performed at 10 ppm [Ca]_{Matrix}. The B/Ca and Sr/Ca data are acid blank corrected;
 43 however, none of the data are corrected for instrument drift. The grey area in figures (A) and
 44 (B) represents $\pm 5\%$ variation from the first analysis. The red symbols represent analyses
 45 performed with a quartz cyclonic spray chamber. All other analyses were performed with a
 46 Teflon[®] Scott type spray chamber – black symbols. The symbols are as follows: 0.1 M HNO₃
 47 – squares with addition symbols; 0.4 M HNO₃ – open addition symbols; 0.1 M HNO₃ +
 48 0.006 M HF – left-triangles; 0.1 M HNO₃ + 0.03 M HF – right-triangles; 0.1 M HNO₃ + 0.06
 49 M HF – circles; 0.1 M HNO₃ + 0.12 M HF – squares; 0.1 M HNO₃ + 0.18 M HF – diamonds;
 50 0.1 M HNO₃ + 0.24 M HF – up-triangles; and 0.1 M HNO₃ + 0.3 M HF – down-triangles.

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